METHOD AND SYSTEM FOR EXTRACTING HYDROCARBON FUEL PRODUCTS FROM WASTE MATERIAL

BACKGROUND OF THE INVENTION

5 1. Field of the Invention

The present invention relates generally to recyclable materials recovery systems, and more specifically, to a method and system for extracting useable hydrocarbon fuel products from waste material by conversion of fuel pre-products to gas.

2. Background of the Invention

A significant amount of waste material produced by residential and commercial facilities comprises plastics. Unless a recovery and/or recycling system is implemented, these plastics ultimately end up in landfills or are incinerated, producing undesirable gaseous pollutant products and ash. Some plastics may be recycled and used in part to form new plastic products, but it is possible to revert plastics to constituent chemical components or other compounds and re-use these compounds to produce fuel or for other manufacturing purposes.

Other waste materials, such as oilfield sludge (a mixture of tar, sand and dirt), absorbent materials that have been used to

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clean up oil spills and byproducts of other manufacturing and refining processes may contain useable hydrocarbon fuel components, also. It would be desirable to be able to process these materials in the same manner as the above-mentioned plastic waste material.

In particular, it is possible to produce a diesel-like fuel from hydrocarbon compounds that may be extracted from plastics and other waste. At high temperatures, the compounds are a gaseous mixture containing various hydrocarbons, aromatics and other gases. The gases may then be further separated and processed by distillation or other refining means to produce various usable fractions. In general, the resulting liquid condensed from the extracted gases cannot be burned in a diesel engine, as the spectrum of hydrocarbons produced from a mixture of plastics, or a solitary plastics contains a high fraction of "hot" components such as octane that will destroy a diesel engine unless the fraction is reduced to tolerable levels via a refining process. In general, all of the product should be useable, as octane can be used to make a gasoline fuel and lighter components may be "cracked" to form propane and synthetic natural gas fuels.

Several existing methods and systems have been proposed to revert plastic materials to gas from which fuel may be produced. In general, these systems fall into two categories: low

temperature vapor extraction methods and high temperature pyrolytic conversion methods. The pyrolytic conversion methods require high energy input and generate gaseous fuel products such as butane and methane which require compression and large volume storage per BTU. The efficiency of conversion is very low, as the long-chain hydrocarbons present in plastics are converted to very short-chain hydrocarbon fuel components, wasting the energy available in the longer chains already present in plastics and other waste material.

Vapor extraction methods in the existing art have a low production throughput and are prone to a build-up of cross-linked polymers that must be removed from the equipment and a build-up of heavy hydrocarbon components that are not effectively removed from the system. They also are susceptible to environmental conditions such as barometric pressure and ambient temperature. These drawbacks have made existing vapor extractio

systems not practical for both economic and production volume reasons.

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Therefore, it is desirable to provide a method and system for extracting usable hydrocarbon fuel products from waste material in an energy efficient manner having high production throughput.

SUMMARY OF THE INVENTION

The above objective of providing efficient and highthroughput extraction of useable hydrocarbon fuel components from waste material is achieved in a method and system. The method and system introduce waste material to a melting chamber having a substantially constant temperature and a liquid fuel pre-product is generated. The liquid is introduced to a process chamber at a substantially constant higher temperature and a negative relative pressure (vacuum) is applied to cause the liquid to off-gas the useable hydrocarbon fuel components in gaseous form. The liquid is agitated as well as heated to promote off-gassing. The melting chamber and process chamber may be a staged feed system having two different portions and an auger may be used to feed the waste material through the chambers. The remainder of the waste material is ejected at the far end of the process chamber and may be combined with the output of other chambers and re-processed for further extraction of useable fuel components.

The foregoing and other objectives, features, and advantages of the invention will be apparent from the following, more particular, description of the preferred embodiment of the invention, as illustrated in the accompanying drawings.

BRIEF DESCRIPTION OF THE DRAWINGS

- $\begin{tabular}{lll} \textbf{Figure 1} is a simplified pictorial diagram depicting a \\ system in accordance with an embodiment of the present invention. \\ \end{tabular}$
- Figure 2 is a detailed pictorial diagram depicting a system in accordance with an embodiment of the present invention.
- Figure 3 is a pictorial diagram depicting details of the processing of plastic material within process chamber 12 of Figure 2.
 - Figure 4A is a pictorial diagram depicting a detailed side view of pressure chamber 19 of Figure 2.
 - Figure 4B is a pictorial diagram depicting a detailed end view of pressure chamber 19 of Figure 2.
- Figure 5 is a pictorial diagram depicting details of the 20 auger unit 27 of Figure 2.
 - Figure 6 is a pictorial diagram depicting a system in accordance with an alternative embodiment of the invention.

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DETAILED DESCRIPTION OF THE EMBODIMENTS

Referring now to the figures and in particular to Figure 1, a simplified depiction of a system for extracting hydrocarbon fuel products from waste material in accordance with an embodiment of the present invention is shown. Waste material is introduced to a hopper 1 and is liquified in a liquification chamber 2. Liquification chamber 2 can accept plastic material, which is generally shredded and compressed. The plastic material is melted to form a liquid 3 by heating ligification chamber 2 to a temperature substantially equal to 585 degrees Fahrnheit. The above-stated temperature is chosen as a minimum temperature to reduce a plastic mixture containing typical recyclable plastics to a liquid and a maximum temperature to avoid forming a crosslinked solid. Cross-linked polymers generally have a higher melting point than recyclable plastic waste and will not readily convert to hydrocarbon fuel components via the system once they have been produced by over-temperature. Therefore, the temperature of liquification chamber 2 should be carefully controlled so that no solid plastics remain and so that none of the plastic material becomes cross-linked.

Liquification chamber 2 is included to ensure that solid plastic material is not introduced to further processing, which will be performed at a higher temperature. The higher temperature

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processing will form cross-linked chains within any solid plastic material. Processing liquified plastics provides a mechanism for avoiding the overheating of solid plastics and thus the crosslinking of the plastics' hydrocarbon chains. A pump 5 moves the liquified plastics 3 to a process chamber 4, where the liquified plastics (which contain hydrocarbon fuel pre-products) are further heated to a temperature substantially equal to 850 degrees Fahrenheit and are permitted to off-gas. The above-stated temperature is a maximum temperature chosen to cause the liquified plastic to off-gas rapidly, while avoiding "cracking" of the bydrocarbon fuel products such as decane to shorter-chain hydrocarbon components, as is done in pyrolytic converters. The higher temperature of process chamber 4 is generally above the flash point of the gaseous fuel components that are being extracted, and therefore the system must substantially prevent the introduction of air within process chamber 4 (and possibly liquification chamber 2). Otherwise the fuel products may combust in the chamber, creating a hazard and wasting the fuel components extracted.

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Some off-gassing will also occur in liquification chamber 2 and these gases may be removed by a vent 8A, but primarily, the gases (which contain hydrocarbon fuel products) will be extracted via a vacuum pump 8. Vacuum pump 8 is included to remove the gases produced by the off-gassing liquid plastic material 7.

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Without vacuum pump 8, insufficient gas will exit the system, and the heavier vapor components (such as parrafin) will remain in the system. Additionally, the effects of ambient barometric pressure and temperature on the system are eliminated. A system without negative relative pressure applied to the process chamber will not remove heavier gaseous hydrocarbon fuel components when the outside temperature falls too low. As a result, continued "cracking" of the hyrdocarbon fuel products occurs and under certain ambient conditions, the output will be only lighter components that cannot condense to form a liquid fuel.

Additionally, the overall output of a non-vacuum driven system under the above-described conditions will be reduced to a small fraction of the potential system production.

Once the gases are removed, they are introduced to a processing system 9 for condensation of heavier fuel components, further refining of heaviest fuel components and potential extraction and storage of lighter fuel components. But, the lighter fuel components are useful for combustion heating of process chamber 4 and liquification chamber 2 and will generally be used for this purpose.

The present invention may also be adapted (or used in an existing form) for the processing of other material containing usable hydrocarbon fuel pre-products. Drill cuttings or other

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waste from oil drilling sites (which is a mixture of sand or dirt and crude oil) may be heated to extract the contained hydrocarbon fuel pre-products and off-gas fuel components. Optionally, liquification chamber may not be required for this type of processing (if the waste material does not contain polymers which will become cross-linked) and systems may be specially adapted in accordance with embodiments of the present invention.

Referring now to Figure 2, a plastic reversion system 10, in accordance with an embodiment of the present invention is shown. A process chamber 12 is formed from a cylindrical pipe with an auger 11 disposed within and passes through within a heating unit 27. A separate liquification chamber is not implemented within system 10 but is provided by heating a first portion of auger 11 and process chamber 12 (before entering heating unit 27) to a lower temperature substantially equal to 585 degrees Fahrenheit for the first section of the process chamber 12 via electric heat sheath 28. Auger 11 is rotated by a drive system 15 at a substantially constant rate. Plastic material chips are introduced to process chamber 12 from a feed hopper 13 and a feed auger 14 may be used to compress the plastic material (and optionally grind the plastic chips from larger portions) for introduction to process chamber 12 under pressure, eliminating introduction of air. Auger 11 drives the plastic material through

gas.

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11 and feed auger 14 may be driven by an electric motor and gearbox combination, a hydraulic motor or pneumatic motor depending on requirements of the particular system application.

Process chamber 12 is generally a metal pipe, and the diameter of auger 11 and process chamber 12 are determined by throughput requirements. The length of auger 11 and angle of the auger flights are chosen to determine the "dwell time" (the time the plastic material take to travel through process chamber 12), which is a critical factor in reverting the plastic material to

Process chamber 12 is heated by a heating system comprising a heater 16 for heating air, ducting 17 for delivering the heated air to ports 18 within heating unit 27 that are coupled directly to a pressure chamber 19 that encloses a portion of process chamber 12. Electic heat sheath 28 may be replaced by a heating system porting air from heater 16 or a second heater, so that the lower liquification temperature may be maintained over the first segment of process chamber 12 that acts as a liquification chamber. Air exits heating unit 27 through ports 20 and is recirculated via a duct 22 and blower 21, returning to heater 16. The heating system produces a high pressure air stream around process chamber 12. The air is heated to raise the temperature of

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the portion of auger 11 and process chamber 12 within heating unit 27 to approximately 850 degrees fahrenheit.

The plastic material undergoes phase changes as it is driven around the heated auger 11, and near the end of auger 11 exit pipes 23 are connected to remove the gaseous mixture produced within process chamber 12 and at the end of auger opposite feed hopper 13. The gas mixture removed at exit pipe 23 is a mixture of many different compounds and gaseous elements and may be condensed, refined or otherwise processed to yield useful products. In general, a very clean-burning fuel product may be refined from the gaseous mixture. Yield output is approximately one gallon of distillate from eight to nine pounds of plastic material. Waste material 25 from the process, called "char" or ash is dropped from a lower exit pipe 24 to a char pit 26 for disposal.

A vacuum system 29 applies a negative relative pressure to process chamber 12, assisted by an "air dam" that is created by compressed plastic material at a point along auger 11 before the plastic material has reached a completely liquid state.

Vacuum system 29 generates a suction that draws against the air dam, generating a negative relative pressure throughout process chamber 12, which aids in the conversion of the plastic material

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from liquid to gas and increases the yield of the system to a practical production level.

Referring now to Figure 3, the processing of plastic material within the system of Figure 2 is depicted with reference to a detailed representation of process chamber 12. Feed auger 14 compresses plastic chips from feed hopper 14 and introduces them to auger 11 within process chamber 12. The rotation of auger 11 moves the plastic material, which is still in solid form, and heat is transferred from heat sheath 28 to auger 11 and process chamber 12 to melt the plastic material.

At the end of length L1, the plastic material has become liquified and pools on the bottom of process chamber. It is within length L1 that the approximate location of air dam lies (prior to complete liquification), and a negative relative pressure will be exerted on the plastic material ahead of this location. In length L2, the plastic material becomes completely liquified and within length L3, the material is converted to gaseous form with some solid waste product remaining (char). At the end of process chamber 12, the char are ejected for disposal or further processing. The gases are removed for processing by vents located within length L3 and coupled to a vacuum system.

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While the above description and illustrations depict a process having generally defined phase boundaries and temperatures and the description suggests uniform composition, in practice, the plastic material introduced to feed hopper 13 may comprise many different plastic materials, and the mixture may vary substantially over time. Therefore, different temperatures may be used and the material within process chamber 12 may transition to liquid form at varying locations. But, it has been found that for a well-mixed feed stock, the behavior of the plastic material is very homogeneous, approximating the behavior of an average material.

Critical to the operation of the system is that the rate of feed and temperature within process chamber 12 be carefully controlled. If the temperature of liquification portion of process chamber 12 is too low, or the rate of feed is too high the plastics will not be completely melted before reaching length L3 and the solid polymer materials will cross-link, forming a very tough product that will not gassify, and may not move through the process chamber at all, jamming system 10, or producing excessive waste at the char output. If the temperature of the liquification portion of process chamber is too high, the material will cross-link rather than melt.

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If the temperature of process chamber 12 within heating unit 27 is too low, or the rate of feed too high, little or no offgassing will occur, producing excessive waste at the char output and a low yield at the gas output. The plastics reverted by system 10 generally begin to off-gas at a temperature higher than 400 degrees fahrenheit, so the operating temperature of heating unit 27 (850°F) ensures that the process temperature will be high enough to revert the plastics, but low enough to avoid cracking of the gases that have been extracted. In general, the rate of feed and temperature may be selected based on the materials being processed, average characteristics and other criteria to maximize throughput, conversion efficiency or both.

Referring now to Figures 4A and 4B, details of pressure chamber section 19A within pressure chamber 19 of Figure 1 are depicted. Figure 3A is a side view of a pressure chamber section 19A (there are three such sections making up pressure chamber 19 of Figure 1). Pressure chamber section 19A comprises an inner pipe 41 and an outer pipe 40. The cylindrical wall of inner pipe 41 is perforated (to avoid the formation of "hot spots") and is connected to outer pipe 40 by a plurality of pipes that protrude through to the outer wall of outer pipe 40, supporting inner pipe 41 along with process chamber 12 (Figure 1). A port 18A is directed at pressure chamber 19 to supply heated air around

pressure chamber 19 to heat process chamber 12. Figure 4B depicts a side view of pressure chamber 19, showing the orientation of pipes 42 and port 18A, as well as the location of perforations 43.

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Referring now to **Figure 5**, details of heating systems within the system of **Figure 2** are shown. Heat sheath **28** is an electric heater coupled to a power system **52**, but may be replaced with a air circulated gas system as used in heating unit **27**.

Heating unit 27 contains ports 18A and 18B that supply heated air to chambers 50, heating pressure chamber sections 19A and thus process chamber 12. Ports 20 return air to the heating system forming a closed loop through chambers 50. Port 18A differs from the other two ports 18B in that the outlet location is disposed slightly further along process chamber 12, providing more heat toward the end of length L1 of Figure 3.

Referring now to **Figure 6** a system in accordance with an alternative embodiment of the invention is shown. In the alternative embodiment depicted, six process chambers **61A** are placed in parallel and all pass through a common heating unit **62**, providing more efficient use of the heating system. The char outputs of process chambers **61A** are combined in a piping manifold **64** and are introduced to a secondary process chamber **61B**, that

extracts any remaining gaseous fuel products. The char output of process chamber 61B is then removed for discard or use. While the illustration shows a flattened parallel construction, in practice, the system may be formed with process chambers 61A arranged so that their cross sections form a ring, or they may be closely packed to improve coupling to heater 62 and reduce the exterior size of heater 62.

While the invention has been particularly shown and described with reference to the preferred embodiments thereof, it will be understood by those skilled in the art that the foregoing and other changes in form, and details may be made therein without departing from the spirit and scope of the invention.